

## Inorganic and Analytical Chemistry (higher education planning materials like oil and natural gas)



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### ***Reviews***



*It is one of my personal favorite pdf. This really is for all those who state there was not a really worth looking at. I realized this book from my dad and i encouraged this pdf to understand.*

***(Katlynn Haag)***

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paperback. Book Condition: New. Ship out in 2 business day, And Fast shipping, Free Tracking number will be provided after the shipment. Pages Number: 387 Publisher: oil industry Pub. Date :2007-08-01 1st edition 1st printing 2007-08-01 This textbook introduces the first distributed system from the macro (dilute solution, gel) and the fundamental nature of the basic principles of chemical reactions (energy changes in reaction rate, the reaction direction, the equilibrium move), and then introduced from the microscopic structure of matter (atoms, molecules, crystals) the basics. Then outlined the basics of quantitative chemical analysis, discusses the various types of chemical equilibrium in solution and titration analysis in the application, and the most commonly used instrument analysis made profile. Finally an important and complex element of the separation and enrichment material. Contents: Introduction Chapter I Section I of gas and solution gas dispersion system II III IV non-electrolyte solution, the concentration of dilute solution of colloidal solution according to the number of V VI VII polymer solution and emulsion section of the electrolyte solution thinking questions the basis of chemical thermodynamics chapter exercises and basic knowledge of chemical equilibrium thermodynamics Section II thermo-chemical reaction in the direction III IV chemical equilibrium problems thinking exercises Section III Chapter chemical reaction chemical reaction rate rate of chemical reaction rate theory the concept of Section III of the factors affecting the rate of chemical reaction problems thinking exercises Chapter Section periodic atomic structure and elements of the hydrogen atom spectrum and Bohr's theory II features the first micro-particle motion three describe the motion state of extranuclear fourth multi-electron atomic structure of elements of V ionization energy, electron affinity and electronegativity thinking questions Exercises Chapter molecular structure and crystal and molecular structure of key parameters of Section nature of the chemical bond theory of Section II...

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